Watkin, D. J., Prout, C. K. Carruthers, R. J. & Betteridge, P. (1996). CRYSTALS. Issue 10. Chemical Crystallography Laboratory, University of Oxford, England.

Watkin, D. J., Prout, C. K. & Lilley, P. M. dc Q. (1994). RC93. Chemical Crystallography Laboratory, University of Oxford, England. Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). CAMERON. Chemical Crystallography Laboratory, University of Oxford, England.

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# *N*-(*p*-Hydroxybenzylidene)phenylamine *N*-Oxide

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#### **Abstract**

The crystal structure of the title compound,  $C_{13}H_{11}NO_2$ , confirms that it exists as a nitrone. The geometry about C=N is Z. The torsion angles O8-N8-C7-H7  $[174.55(2)^{\circ}]$  and C8—N8—C7—H7  $[-0.89(4)^{\circ}]$ indicate the trans and cis positions around the nitrone moiety.

## Comment

The title compound, (I), has been analysed as part of crystallographic studies on substituted N-phenylnitrones.

$$HO - N^{+} - N^{-}$$

The molecule exists as a nitrone. The C=N distance of 1.304 (3) Å is within  $3\sigma$  of 1.279 (8) Å reported by Allen et al. (1987). The dihedral angle between the C8. N8, C7 and N8, C7, C1 planes is 2.61 (2)°. The nitrone group formed by atoms C8, N8, C7 and C1 attached to two phenyl rings is nearly planar. Atoms N8, C8, C7 and C1 deviate from the least-squares plane by -0.008(2), -0.013(2), -0.017(2) and -0.015(2) Å, respectively. The C1—C7 distance of 1.440(3) Å is nearly equal to the value of 1.438(4) Å reported by Bedford, Chaloner & Hitchcock (1991). The phenyl bond lengths and angles are in agreement with the expected values reported by Allen et al. (1987). The dihedral angle between the two phenyl rings is 55.3 (1)°.

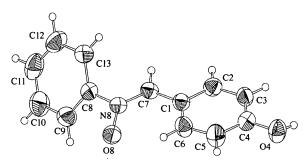


Fig. 1. The molecular structure of (I) showing 50% probability displacement ellipsoids.

# **Experimental**

A solution of 4-hydroxybenzaldehyde (1.22 g) in ethanol (25 ml) was added dropwise with stirring to a solution of Nphenylhydroxylamine (1.09 g) in ethanol (10 ml). The mixture was warmed for about 1 h at 333 K. On cooling, crude crystals of the title compound separated and were recrystallized from ethanol [m.p. 487 K; yield 1.76 g (83%)]. The title compound was prepared in the Department of Industrial Chemistry, Alagappa University, Karaikudi, India.

Crystal data

•	
$C_{13}H_{11}NO_2$	Mo $K\alpha$ radiation
$M_r = 213.23$	$\lambda = 0.7107 \text{ Å}$
Orthorhombic	Cell parameters from 25
Pbca	reflections
a = 18.033(9)  Å	$\theta = 5-25^{\circ}$
b = 14.019(9)  Å	$\mu = 0.085 \text{ mm}^{-1}$
c = 8.973(5)  Å	T = 293  K
$V = 2269 (2) \text{ Å}^3$	Rectangular
Z = 8	$0.25 \times 0.20 \times 0.10 \text{ mm}$
$D_x = 1.249 \text{ Mg m}^{-3}$	Yellow
$D_m$ not measured	

Data collection

Rigaku AFC-7S diffractom-	$R_{\rm int} = 0.032$
eter	$\theta_{\text{max}} = 24.87^{\circ}$
$\omega$ –2 $\theta$ scans	$h = 0 \rightarrow 21$
Absorption correction:	$k = 0 \rightarrow 16$
$\psi$ scan (North, Phillips	$l = -10 \rightarrow 0$
& Mathews, 1968)	3 standard reflections
$T_{\min} = 0.922, T_{\max} = 1.000$	every 100 reflections
1875 measured reflections	frequency: 150 min
1874 independent reflections	intensity decay:
843 reflections with	negligible
$I > 2\sigma(I)$	

Refinement

Refinement on 
$$F^2$$
  $(\Delta/\sigma)_{\rm max} = 0.048$   $R[F^2 > 2\sigma(F^2)] = 0.035$   $\Delta\rho_{\rm max} = 0.127 \ {\rm e \ \mathring{A}^{-3}}$   $\Delta\rho_{\rm min} = -0.117 \ {\rm e \ \mathring{A}^{-3}}$ 

S = 0.9721874 reflections 145 parameters H atoms not refined  $w = 1/[\sigma^2(F_o^2) + (0.0407P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$  Extinction correction: none Scattering factors from International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

O4—C4	1.353(3)	C1—C6	1.404 (3)
C4—C3	1.386(3)	C5—C6	1.370(3)
C4C5	1.390(4)	C13—C12	1.373 (4)
C8—C9	1.376 (4)	C9-C10	1.385 (4)
C8—C13	1.376 (4)	C12—C11	1.378 (5)
C2—C3	1.370(4)	C11—C10	1.368 (5)
C2—C1	1.401 (4)		
C7—N8—C8	120.6(2)	C9—C8—C13	121.3 (3)
O8-N8-C8	116.3(2)	C9—C8—N8	120.7 (2)
N8—C7—C1	127.3(2)	C13—C8—N8	118.0(2)
O8-N8-C7-C1	1.8 (4)	N8C7C6	167.1 (2)
C8—N8—C7—C1	-177.4(2)	N8C1C2	-12.8(4)
C7-N8-C8-C9	-43.2(3)	O4—C4—C5—C6	-178.0(2)
O8-N8-C8-C9	141.0(2)	N8—C8—C13—C12	-178.2(2)
C7—N8—C8—C13	137.3 (2)	N8—C8—C9—C10	178.7 (2)
O8—N8—C8—C13	-38.6(3)	O8—N8—C7—H7	174.55 (2)
O4—C4—C3—C2	179.8 (3)	C8—N8—C7—H7	-0.89(4)

The title structure was solved by direct methods using SHELXS86 (Sheldrick, 1985). All non-H atoms were refined with anisotropic displacement parameters. All H atoms were located from difference Fourier maps and were included in the structure-factor calculations with isotropic displacement parameters equal to  $1.1U_{eq}$  of the respective carrier atom, but the parameters were not refined (SHELXL93; Sheldrick, 1993). The geometrical calculations were performed using PARST (Nardelli, 1983).

Data collection: TEXSAN (Molecular Structure Corporation, 1989). Cell refinement: TEXSAN. Data reduction: TEXSAN. Molecular graphics: ORTEPII (Johnson, 1976). Software used to prepare material for publication: SHELXL93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1054). Services for accessing these data are described at the back of the journal.

### References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-S19.
Bedford, R. B., Chaloner, P. A. & Hitchcock, P. B. (1991). Acta Cryst. C47, 2484-2485.

Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Molecular Structure Corporation (1989). TEXSAN. Single Crystal Structure Analysis Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Nardelli, M. (1983). Comput. Chem. 7, 95-98.

North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.

Sheldrick, G. M. (1985). SHELXS86. Program for the Solution of Crystal Structures. University of Göttingen, Germany.

Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.

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# 4-Chlorophenyl 2,6-Dimethylphenyl Sulfone

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#### Abstract

The phenyl rings in molecules of the title compound,  $C_{14}H_{13}ClO_2S$ , are perpendicular to one another. The crystal is stabilized by weak C—H···O intermolecular hydrogen bonding.

#### Comment

Like many sulfonamides and sulfones, the title compound, (I), has medicinal value (Krishnaiah, Narayana Raju, Lu, Chen & Narasinga Rao, 1995). It displays bond lengths which agree well with accepted values [International Tables for Crystallography (1992, Vol. C, Table 9.5.1.1)]. The C1—S1—C1' angle of 103.2 (1)° is markedly less than the tetrahedral angle, 109.5°. This may be due to the short O2···C7 contact of 2.730 (4) Å. The phenyl rings are perpendicular [dihedral angle 91.2 (1)°] to one another.

The crystal packing is stabilized by a weak intermolecular hydrogen bond [C2'—H2' 0.94 (2), C2'···O2<sup>i</sup> 3.452 (4), H2'···O2<sup>i</sup> 2.72 (3) Å and C2'—H2'···O2<sup>i</sup> 136 (2)°; symmetry code: (i)  $-x + \frac{1}{2}$ ,  $y - \frac{1}{2}$ ,  $z + \frac{3}{2}$ ].

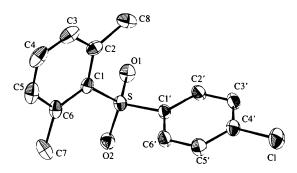


Fig. 1. The 50% probability displacement ellipsoid plot of the title molecule with the atom-numbering scheme. H atoms have been omitted.